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Glycal Ethers as Starting Materials in C-Nucleoside Synthesis

Andrew Kaye^a; Colin B. Reese^a; Stephen Neidle^b

^a Department of Chemistry, King's College London, London, England ^b Institute of Cancer Research, Surrey

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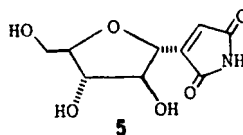
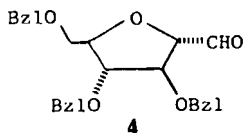
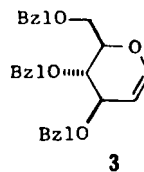
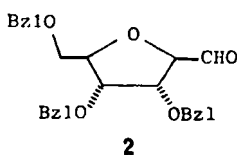
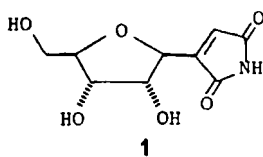
GLYCAL ETHERS AS STARTING MATERIALS IN
C-NUCLEOSIDE SYNTHESIS

Andrew Kaye and Colin B. Reese*, Department of Chemistry,
King's College London, Strand, London WC2R 2LS, England.

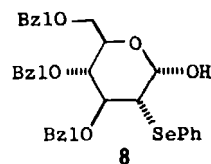
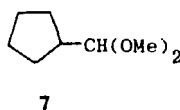
Stephen Neidle, Institute of Cancer Research, Clifton
Avenue, Sutton, Surrey, SM2 5PX.

Abstract. When 3,4,6-tri-O-benzyl-D-glucal (3) is treated with thallium (III) nitrate, trihydrate in acetonitrile, 2,5-anhydro-3,4,6-tri-O-benzyl-D-mannose (4), a precursor of the showdomycin analogue (5), is obtained. Ring-contraction of (3) can also be effected by a three-step process [(i) PhSeCl, tetrahydrofuran, (ii) Et₃N, H₂O; (iii) 3-ClC₆H₄CO₂H, MeOH]. Other examples of these ring-contraction reactions are described.

Trummlitz and Moffatt have reported¹ a synthesis of the nucleoside antibiotic showdomycin (1), based on 2,5-anhydro-3,4,6-tri-O-benzyl-D-allose (2). Although the latter showdomycin synthesis is elegant, the procedure used¹ for the preparation of (2) was rather lengthy. It occurred to us that it should be possible to prepare analogues of (2), and hence analogues of showdomycin, by the ring-contraction of glycal benzyl ethers.

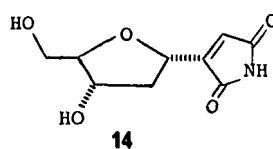
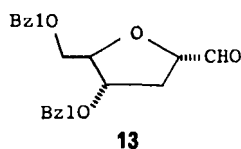
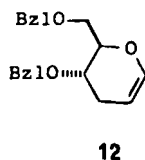
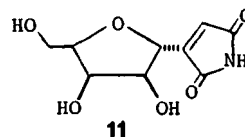
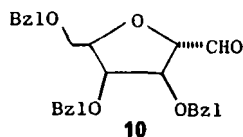
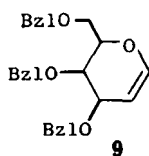


Treatment of 3,4,6-tri-O-benzyl-D-glucal² (3) with a stoichiometric quantity of thallium (III) nitrate, trihydrate³ (TTN) in acetonitrile solution at room temperature for 1 hr gave⁴ what is assumed to be 2,5-anhydro-3,4,6-tri-O-benzyl-D-mannose (4) as the main product. The latter crude aldehyde (4) was converted^{4,5} into the showdomycin analogue (5), m.p. 134°C, by the six-step procedure used by Trummelitz and Moffatt¹ to convert (2) into (1). The overall yield of (5) for the seven steps, starting from (3), was ca. 18%.



An alternative procedure for ring-contraction was suggested by the recent report⁶ that treatment of trans-2-methoxybenzeneselenenocyclohexane (6) with m-chloroperbenzoic acid in methanol gave 2-(dimethoxymethyl)tetrahydrofuran (7). When 3,4,6-tri-O-benzyl-D-glucal (3) was allowed to react with a slight excess of benzeneselenenyl chloride in dry tetrahydrofuran at room temperature and the products then treated with triethylamine and water, the benzeneselenenol adduct (8) was obtained⁷ and isolated as a crystalline solid, m.p. 98°C, in 67% yield. The configuration at the site of selenenation [i.e. at C(2)], the assignment of which is based on ¹H-n.m.r. spectroscopic data, differs from that reported in a related study by Sinay⁸ and his coworkers⁸. Treatment of (8) with slightly more than two mol. equiv. of m-chloroperbenzoic acid in methanol at room temperature for 30 min gave the crude aldehyde (4), which was again converted⁷ into the α -anomer of ara-showdomycin (5) by Trummelitz and Moffatt's procedure¹. The overall yield of (5) for the seven steps, starting from (8), was 19.5%.

The reaction between 3,4,6-tri-O-benzyl-D-galactal (9) and TTN proceeded slowly and a complex mixture of products was obtained⁷. However, when (9) was treated first with benzeneselenenyl chloride and then with triethylamine and water, and the resulting benzeneselenenol adduct oxidized with m-chloroperbenzoic acid in methanol, a crude aldehyde-compound with putative structure (10) was obtained⁷. The latter product (10) was readily converted⁷ into (11), the α -anomer of lyxo-showdomycin, m.p. 144 - 145°C. The overall yield of (11) was 13% for the nine steps starting from (9). Finally, when 4,6-di-O-benzyl-



3-deoxy-D-glucal (12) was treated⁴ with TTN in acetonitrile for 30 min at room temperature, the reaction proceeded smoothly and the putative aldehyde (13) was obtained and isolated as its N,N'-diphenylimidazolidine derivative⁹, m.p. 47 - 49°C, in 61% yield based on (12). The latter derivative was converted⁴ by Trummlitz and Moffatt's procedure¹ into α-2'-deoxyshowdomycin (14), m.p. 112°C. The overall yield of (14) for the nine steps, starting from (12), was ca. 14.5%. The structures of all three showdomycin analogues [(5), (11) and (14)] described above have been confirmed by X-ray crystal structure determinations^{4,7}.

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REFERENCES

1. Trummlitz, G.; Moffatt, J.G. J. Org. Chem. 1973, **38**, 1841.
2. Blackburne, I.D.; Fredericks, P.M.; Guthrie, R. D. Aust. J. Chem. 1976, **29**, 381.
3. McKillop, A.; Hunt, J.D.; Kienzle, F.; Bigham, E.; Taylor, E.C. J. Am. Chem. Soc. 1973, **95**, 3635.
4. Kaye, A.; Reese, C.B.; Neidle, S. Tetrahedron Lett. 1988, **29**, 1841.
5. Before recrystallisation, (5) appeared to be contaminated with ca. 5 - 10% of an impurity which was probably its β-anomer.
6. Uemura, S.; Fukusawa, S.; Toshimitsu, A. J. Chem. Soc., Chem. Commun. 1983, 1501.
7. Kaye, A.; Reese, C.B.; Neidle S., submitted for publication.

8. Jaurand, G., Beau, J.-M.; Sinaÿ, P. J. Chem. Soc., Chem. Commun. 1981, 572.
9. Albrecht, H.P.; Repka, D.B.; Moffatt, J.G. J. Org. Chem. 1973, 38, 1836.